# PRESENTATION OF THE EXPERIENCE





# Practical work : What's the best way to heat water?



#### 1) Hotplate's energy efficiency

At First, we need power value which is on the hotplate to know how much energy the hotplate is spending (power unit is the Watt).

P = .....W

#### The experimentation:

Heat the hotplate for a few minutes at maximum power.

Meanwhile pour into a flask 200 ml of water, using a graduated cylinder.

Note the initial temperature (It) of the cold water by using a thermometer:

lt =

Heat the 200 ml of water during exactly 5 minutes at maximum power.

After 5 minutes, remove the flask from the hotplate.

Shake before raising the temperature of the hot water.

Ft =

Note the final temperature (Ft) of the hot water by Using the thermometer:

#### With these results, we can find hotplate's efficiency!

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Calculation of electrical energy consumed during 5min:

Reminder: E_{electric} = P (power) x t (time) with E in watt.min

E_{electric} =

Give the value of E_{electric} in Joules (x 60)
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Calculate water's thermal energy E<sub>thermal</sub>

# **<u>Reminder</u>** : $E_{thermal} = w$ (weight) x c (capacity = 4180 J.kg<sup>-1</sup> .K<sup>-1</sup>) x (Ft-It) $E_{thermal} =$ Finally, we can calculate hotplate's efficiency! $Ef_1 = (E_{thermal} / E_{electric}) \times 100$

### Conclusion

## 2) Heating mantle's energy efficiency

Now, we'll do the same experimentation but with a heating mantle:

Power value	
(watt)	W
Initial temp. (It)	°C
Final temp. (Ft)	°C

#### With these results, we can find hotplate's efficiency!

Calculation of electrical energy consumed during 5min: **Reminder** :  $E_{electric} = P (power) \times t(time)$  with E in watt.min Give the value of  $E_{electric}$  in Joules

Calculate water's thermal energy E<sub>thermal</sub> Reminder : E<sub>thermal</sub> = w x c (= 4180 J.kg<sup>-1</sup> .K<sup>-1</sup>) x (Ft-It)

Finally, we can calculate heating mantle's efficiency!

 $Ef_2 = (E_{thermal} / E_{electric}) \times 100$ 

#### Conclusion











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